MATERIAL SAFETY DATA SHEET

	DATA SHEET
PRODUCT NAME	CAS#
Hydrogen Chloride	7647-01-0
TRADE NAME AND SYNONYMS	DOT I.D. NO.
Hydrogen Chloride	UN 1050
CHEMICAL NAME AND SYNONYMS	DOT HAZARD CLASS
Hydrochloric Acid, Anhydrous	Division 2.3
ISSUE DATE AND REVISIONS	FORMULA
Revised March 2007	HCl

HEALTH HAZARD DATA

EMERGENCY OVERVIEW

Hydrogen Chloride is a corrosive high pressure liquid and gas. Hydrogen Chloride may cause liver damage. Also, it may cause eye, skin, and respiratory tract burns.

SYMPTOMS OF EXPOSURE

<u>Inhalation</u>: Corrosive and irritating to the upper and lower respiratory tracts. It hydrolyzes very rapidly yielding hydrochloric acid. This is caused from exposure to volatile inorganic acids. Symptoms include tearing of eyes, cough, labored breathing and excessive salivary and sputum formation. Excessive irritation of the lungs causes acute pneumonitis and pulmonary edema that could be fatal.

Skin Contact: Corrosive, causing severe irritation with chemical burns and ulceration of the skin exhibiting pain, redness, swelling, and early necrosis.

Eye Contact: Immediate pain and irritation is followed by excessive watering and closure of the eyelids can cause redness, irritation of the conjunctiva and possible blindness.

TOXICOLOGICAL PROPERTIES

Hydrogen Chloride is irritating and corrosive to all living tissue. Toxic level exposure to dermal tissue causes hydrochloric acid burns and skin lesions resulting in early necrosis and scarring. Chemical pneumonitis and pulmonary edema result from exposure to the lower respiratory tract and deep lung. Residual pulmonary malfunction might also occur. Burns to the eye may result in lesions and possible loss of vision.

The LC 50 (ppm) is 40,989 ppm for a 5 min exposure for a rat.

The LC 50 (ppm) is 3,124 ppm for a 60 min exposure for a rat.

RECOMMENDED FIRST AID TREATMENT

PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO HYDROGEN CHLORIDE. RESCUERS SHOULD BE EQUIPPED WITH ADEQUATE PERSONAL PROTECTIVE APPARATUS.

<u>Inhalation</u>: Conscious person should be assisted to an uncontaminated area and inhale fresh air. Unconscious persons should be moved to an uncontaminated area and given mouth-to-mouth resuscitation and supplemental oxygen. Keep the victim warm and quiet. Assure that mucus or vomited material does not obstruct the airway by positional drainage. Delayed pulmonary edema may occur. Keep patient under medical observation for at least 24 hours.

<u>Eye Contact</u>: PERSONS WITH POTENTIAL EXPOSURE TO HYDROGEN CHLORIDE SHOULD NOT WEAR CONTACT LENSES. Flush contaminated eye(s) with copious quantities of water for minimum of 15 minutes. Part eyelids with fingers to assure complete flushing.

Skin Contact: Flush affected area with copious quantities of water. Remove affected clothing as rapidly as possible.

HAZARDOUS MIXTURES OF OTHER LIQUIDS, SOLIDS, OR GASES

Reacts hazardously with fluorine, calcium carbide, cesium carbide, rubidium carbide and lithium silicide. Moist Hydrogen Chloride reacts with most metals in a corrosive manner liberating flammable hydrogen gas. It reacts with many organic materials with the liberation of heat.

PHYSICAL DATA			
BOILING POINT	CRITICAL TEMPERATURE		
-85.1°C	51.4°C		
MOLAR SPECIFIC HEAT (25 oC, 1 bar abs, contact volume)	CRITICAL PRESSURE		
20.976J/molºK	82.58 bar abs		
SOLUBILITY IN WATER	SPECIFIC VOLUME(21.1 oC, 1 bar abs)		
Soluble	661.7dm ³ /kg		
EVAPORATION RATE	SPECIFIC GRAVITY (AIR=1)		
N/A	1.27 at 70°F		
APPEARANCE AND ODOR			
Colorless gas with a sharp, pungent odor. Forms dense cloudon contact with the moisture in air.			

FIRE AND EXPLOSION HAZARD DATA

FLASH POINT (Method used)	AUTO IGNITION TEMPERATURE		FLAMMABLE LIMITS % BY VOLUME	
N/A	N/A		lel N/A	uel N/A
EXTINGUISHING MEDIA	-	SPECIAL FIR	RE FIGHTING PH	ROCEDURES
Nonflammable		N/A		
UNUSUAL FIRE AND EXPLOSION HAZARDS				
Reaction of HCl with common metals will form flammable hydrogen gas.				

REACTIVITY DATA

STABILITY		CONDITIONS TO AVOID			
Unstable					
Stable	Х	N/A			
INCOMPATIBILITY (Materials to avoid)					
Water, organic materials.					
HAZARDOUS POLYMERIZA	TION	HAZARDOUS DECOMPOSITION PRODUCTS			
May Occur					
Will Not Occur	Х	Hydrochloric acid on hydrolysis.			

SPILL OR LEAK PROCEDURES

STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED

Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in cylinder or cylinder valve, contact HSG for special advice.

WASTE DISPOSAL METHOD

Waste disposal must be in accordance with appropriate Federal, State, and local regulations. For emergency disposal assistance, contact HSG for specific advice.

SPECIAL PROTECTION INFORMATION

RESPIRTORY PROTECTION (Specify type)				
Positive pressure air line with mask or self-contained breathing apparatus should be available for				
emergency use.				
VENTILATION	SPECIAL			
Hood with forced ventilation.	N/A			
MECHANICAL (Gen.)	OTHER			
N/A	N/A			
LOCAL EXHAUST				
To prevent accumulation above the Ceiling Limit for HCl.				
PROTECTIVE GLOVES				
Nitrile or Neoprene supported				
EYE PROTECTION				
Safety goggles or glasses				
OTHER PROTECTIVE EQUIPMENT				
Safety shoes, safety shower, eyewash "fountain", face shield, and other protective clothing when				
needed.				

SPECIAL PRECAUTIONS*

SPECIAL LABELING INFORMATION	
DOT Shipping Name: Hydrogen Chloride	DOT Hazard Class: Division 2.3
DOT Shipping Label: Poison Gas and Corrosive	I.D. No.: UN 1050
SPECIAL HANDLING RECOMMENDATIONS	

Use only in well-ventilated areas. Valve protection caps must remain in place unless cylinder is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure piping or system. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder.

SPECIAL STORAGE RECOMMENDATIONS

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area away from heavy traffic areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 130°F (54°C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in-first out" inventory system to prevent full cylinders being stored for excessive periods of time.

SPECIAL PACKAGING RECOMMENDATIONS

Most metals corrode rapidly with wet Hydrogen Chloride.

OTHER RECOMMENDATIONS OR PRECAUTIONS

Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with his (written) consent is a violation of Law.

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